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<p>(54) Title: COMPOSITION FOR SUSTAINED RELEASE OF NON-AGGREGATED ERYTHROPOIETIN</p> <p>(57) Abstract</p> <p>A composition, and methods of forming and using said composition, for the sustained release of non-aggregated, biologically active, erythropoietin (EPO). The sustained release composition of this invention comprises a polymeric matrix of a biocompatible polymer and particles of biologically active, aggregation-stabilized EPO, wherein said particles are dispersed within the biocompatible polymer. The method of the invention for producing a composition for the sustained release of biologically active EPO, includes dissolving a biocompatible polymer in a polymer solvent to form a polymer solution, dispersing particles of biologically active, aggregation-stabilized EPO in the polymer solution, and then solidifying the polymer to form a polymeric matrix containing a dispersion of said EPO particles. The method for using a composition of the invention is a method for providing a therapeutically effective blood level of biologically active, non-aggregated erythropoietin in a subject for a sustained period. In this method, a subject is administered an effective dose of the sustained release composition of the present invention.</p>		

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COMPOSITION FOR SUSTAINED RELEASE OF
NON-AGGREGATED ERYTHROPOIETIN

Background of the Invention

Erythropoietin (EPO) is a glycoprotein, which can be
5 isolated from urine or produced by recombinant genetic
engineering, which is used as a hematinic to increase red
blood cell count in a recipient.

Normal production of red blood cells requires the
secretion of EPO by the kidneys. The EPO triggers
10 proliferation and differentiation of a population of
receptive stem cells in the bone marrow, stimulates
hemoglobin synthesis in maturing erythroid cells, and
accelerates release of red blood cells from the bone marrow
into circulation, thereby increasing the red blood cell
15 mass.

EPO is typically used to treat patients with anemia,
particularly wherein the anemia is associated with renal
failure. Currently, aqueous EPO is administered as a
subcutaneous or intravenous bolus three times a week to
20 patients to maintain suitable serum levels of EPO.

For patients chronically receiving EPO, the frequent
injections result in significant variations in serum EPO
levels, as well as patient compliance problems.

To resolve the problems associated with repetitive
25 injections of aqueous EPO, attempts have been made to
formulate controlled release devices containing higher
doses of EPO than a bolus injection, encapsulated within a
polymer and/or protein, wherein the EPO would be released
in vivo over a period of about a week or more.

30 However, these controlled release devices typically
exhibited high initial bursts of EPO release and minimal
EPO release thereafter. Further, due to the high
concentration of EPO within these controlled release

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devices, the EPO molecules (monomer) have tended to aggregate after several days to form aggregated EPO which, unlike EPO monomer, is immunogenic in vivo.

Therefore, a need exists for a means for sustaining
5 the release of biologically active EPO in vivo without causing an immune system response over the release period of the EPO.

Summary of the Invention

This invention relates to a composition, and methods
10 of forming and using said composition, for the sustained release of non-aggregated, biologically active erythropoietin (EPO). The sustained release composition of this invention comprises a polymeric matrix of a biocompatible polymer and particles of biologically active,
15 aggregation-stabilized EPO, wherein said particles are dispersed within the biocompatible polymer.

The method of the invention for forming a composition for the sustained release of non-aggregated EPO, includes dissolving a biocompatible polymer in a polymer solvent to
20 form a polymer solution, dispersing particles of biologically active, aggregation-stabilized EPO in the polymer solution, and then solidifying the polymer to form a polymeric matrix containing a dispersion of said EPO.

The advantages of this sustained release formulation
25 for EPO include longer, more consistent in vivo blood levels of EPO, lower initial bursts of EPO, and increased therapeutic benefits by eliminating fluctuations in serum EPO levels. The advantages also include increased patient compliance and acceptance by reducing the required number
30 of injections. The advantages further include the ability to use smaller amounts of EPO compared to bolus injection regimen because serum EPO levels are maintained closer to therapeutical thresholds.

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Brief Description of the Drawings

Figure 1 is a plot of a) the cumulative release of monomeric EPO, b) the cumulative release of EPO (monomer EPO plus aggregated EPO), and c) the percentage of EPO which is released as a monomer during the interval between an indicated time point and the immediately preceding time point, in vitro in HEPES buffer, from microspheres of unblocked poly(lactide-co-glycolide) polymer (PLGA) (10,000 Dalton MW), containing 10% (w/w) MgCO_3 and 5% (w/w) of the Am1 formulation of Example 1, versus time over a 28 day interval.

Figure 2 is a plot of a) the cumulative release of monomeric EPO, b) the cumulative release of EPO (monomer or aggregate), and c) the percentage of EPO which is released as a monomer, in vitro in HEPES buffer, from microspheres unblocked PLGA (10,000 Dalton MW), containing 10% (w/w) MgCO_3 and 5% (w/w) of the Am7 formulation of Example 1, versus time over a 28 day interval.

Figure 3 is a plot of a) the cumulative release of monomeric EPO, b) the cumulative release of EPO (monomer or aggregate), and c) the percentage of EPO which is released as a monomer, in vitro in HEPES buffer, from microspheres of blocked PLGA (10,000 Dalton MW), containing 10% (w/w) ZnCO_3 and 10% (w/w) of the Zn1 formulation of Example 1, versus time over a 28 day interval.

Figure 4 is a plot of the percent reticulocytes in blood of CS/HC treated and untreated rats, which were subcutaneously injected with 10,000 units of the EPO sustained release microspheres RMAM7, described in Example 5, on day 0 a bolus of 2,000 units of aqueous EPO, administered on day 28, versus time over a 36 day interval.

Figure 5 is a plot of the serum concentration (IU/ml) of EPO in rats, which were subcutaneously administered various EPO sustained release microspheres, described in Example 3, versus time over a 22 day interval.

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Figure 6 is a plot of the percent reticulocytes in blood of rats, which were subcutaneously injected with 10,000 units of various EPO sustained release microspheres, described in Example 3, versus time over a 28 day interval.

5 Figure 7 is a plot of the serum concentration (IU/ml) of EPO in rats, which were subcutaneously injected with 10,000 units of microspheres of Example 2, containing the AM7 formulation of Example 1 in unblocked PLGA polymers having molecular weights of a) 10,000 Daltons, b) 31,000
10 Daltons or c) 48,000 Daltons, versus time over a 28 day interval.

Detailed Description of the Invention

Erythropoietin (EPO), as defined herein, includes all forms of EPO, such as EPO- α and EPO- β . EPO can be derived
15 from animal sources or recombinantly produced as described in U.S. Patent No. 4,703,008.

The EPO used in this invention is biologically active EPO in its molecular (monomeric or non-aggregated) form. Monomeric EPO is typically non-immunogenic.

20 EPO molecules which aggregate may not be biologically active in stimulating red blood cell production. Furthermore, aggregated EPO may induce an immune response resulting in antibodies formed against EPO. This may compromise the efficacy of long-term EPO therapy.
25 Additionally, aggregated EPO may stimulate an auto-immune response to endogenous EPO.

A sustained release of biologically active, non-aggregated erythropoietin is a release which results in measurable serum levels of biologically active, monomeric
30 EPO over a period longer than that obtained following direct administration of aqueous EPO. It is preferred that a sustained release be a release of EPO for a period of about a week or more, and more preferably for a period of about two weeks or more.

35 A sustained release of biologically active, non-aggregated EPO from a polymeric matrix can be continuous or

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non-continuous release with relatively constant or varying rates of release. The continuity of EPO released and level of EPO released can be established by using, *inter alia*, one or more types of polymer compositions, EPO loadings, and/or selection of excipients to produce the desired effect.

Aggregation-stabilized erythropoietin, as defined herein, comprises biologically active, monomeric EPO which is aggregation-stabilized by at least one anti-aggregation agent. In one embodiment, a class of materials, or combination of materials, suitable as anti-aggregation agents includes materials which reduce the solubility of the EPO in aqueous fluids, such as PBS, HEPES or bodily fluids (e.g., lymph), to maintain a localized concentration of EPO below the concentration at which significant aggregation of EPO molecules occurs. A localized concentration of EPO, as defined herein, is the concentration of solvated EPO within, between or immediately surrounding the sustained release microparticles, or device.

In another embodiment, suitable anti-aggregation agent include carbohydrates which prevent significant aggregation of EPO monomer for undetermined reasons.

Significant aggregation is defined as an amount of aggregation resulting in aggregation of about 10% or more of the initial amount of encapsulated EPO monomer. Preferably, aggregation is maintained below about 5% of the initial dose of EPO monomer. More preferably, aggregation is maintained below about 2% of the initial dose. Further discussion of the levels of aggregation observed for the sustained release composition of the present invention is provided in Examples 3 and 4.

In one embodiment, an anti-aggregation agent reduces EPO solubility by precipitating the EPO from the aqueous solution, thereby maintaining a suitably low localized EPO concentration. Suitable materials for precipitating a protein, without denaturing the protein, include salts

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which are in the Hofmeister series of precipitants of serum globulins (or "salting-out salts") as described in by Thomas E. Creighton in *Proteins: Structures and Molecular Principles*, p149-150 (published by W.H. Freeman and Company, New York). Suitable salting-out salts for use in this invention include, for example, salts containing one or more of the cations Mg^{+2} , Li^{+} , Na^{+} , K^{+} and NH_4^{+} ; and also contain one or more of the anions SO_4^{-2} , HPO_4^{-2} , acetate, citrate, tartrate, Cl^{-} , NO_3^{-} , ClO_3^{-} , I^{-} , ClO_4^{-} and SCN^{-} .

10 In another embodiment, an anti-aggregation agent comprises at least one carbohydrate, such as mannitol.

The suitability of candidate anti-aggregation agents for stabilizing EPO against aggregation can be determined by one of ordinary skill in the art by performing a variety of stability indicating techniques such as SEC, polyacrylamide gel electrophoresis (PAGE) and potency tests on protein obtained from EPO particles containing the anti-aggregation agent and for the duration of release from the sustained release composition, as described in Example 3.

20 Suitable particles of biologically active, aggregation-stabilized erythropoietin are solid particles, including lyophilized particles, freeze-dried particles, pressed pellets, and particles formed by any other means known in the art for forming a solid particle from a mixture of two components (e.g., EPO and an anti-aggregation agent) wherein one component is temperature sensitive.

Wherein the biologically active, aggregation-stabilized EPO is lyophilized, it is preferred that said EPO also contain a buffer to maintain pH in a range which will prevent a significant loss of biological activity resulting from pH changes during lyophilization. Suitable pH conditions typically include pH values between about 4.0 and about 8.0. A preferred pH range is between about 5.0 and about 7.0. Suitable pH conditions are typically achieved through use of an aqueous buffer, such as sodium bicarbonate. Examples of preferred buffers include

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phosphate buffers, citrate buffers and combinations thereof.

The EPO particle can also contain other excipients, such as stabilizers and bulking agents. Stabilizers are added to maintain the potency of the EPO over the duration of EPO release. Suitable stabilizers include, for example, carbohydrates, amino acids, fatty acids and surfactants and are known to those skilled in the art. The amount of stabilizer used is based on ratio to the EPO on a weight basis. For amino acids, fatty acids and carbohydrates, such as sucrose, maltose, inulin, dextran and heparin, the mass ratio of carbohydrate to EPO is typically between about 1:1 and about 20:1. For surfactants, such as TWEENTM and PLURONICTM, the mass ratio of surfactant to EPO is typically between about 0.01:1 and about 1:1.

Bulking agents typically comprise inert materials. Suitable bulking agents are known to those skilled in the art.

Polymers suitable to form a polymeric matrix of the sustained release composition of this invention are biocompatible polymers which can be either a biodegradable or non-biodegradable polymer, or blends or copolymers thereof.

Biodegradable, as defined herein, means the composition will degrade or erode in vivo to form smaller chemical species. Degradation can result, for example, by enzymatic, chemical and physical processes. Suitable biocompatible, biodegradable polymers include, for example, poly(lactides), poly(glycolides), poly(lactide-co-glycolides), poly(lactic acid)s, poly(glycolic acid)s, poly(lactic acid-co-glycolic acid)s, polycaprolactone, polycarbonates, polyesteramides, polyanhydrides, poly(amino acids), polyorthoesters, polycyanoacrylates, poly(p-dioxanone), poly(alkylene oxalate)s, biodegradable polyurethanes, blends and copolymers thereof. Biocompatible, non-biodegradable polymers suitable for the modulated release composition of this invention include

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non-biodegradable polymers selected from the group consisting of polyacrylates, polymers of ethylene-vinyl acetates and other acyl substituted cellulose acetates, non-degradable polyurethanes, polystyrenes, polyvinyl chloride, polyvinyl fluoride, poly(vinyl imidazole), chlorosulphonate polyolefins, polyethylene oxide, blends and copolymers thereof.

A polymer, or polymeric matrix, is biocompatible if the polymer, and any degradation products of the polymer, are non-toxic to the recipient and also present no significant deleterious or untoward effects on the recipient's body, such as an immunological reaction at the injection site.

Further, the terminal functionalities of a polymer can be modified. For example, polyesters can be blocked, unblocked or a blend of blocked and unblocked polymers. A blocked polymer is as classically defined in the art, specifically having blocked carboxyl end groups. Generally, the blocking group is derived from the initiator of the polymerization and is typically an alkyl group. An unblocked polymer is as classically defined in the art, specifically having free carboxyl end groups.

Acceptable molecular weights for polymers used in this invention can be determined by a person of ordinary skill in the art taking into consideration factors such as the desired polymer degradation rate, physical properties such as mechanical strength, and rate of dissolution of polymer in solvent. Typically, an acceptable range of molecular weights is of about 2,000 Daltons to about 2,000,000 Daltons. In a preferred embodiment, the polymer is a biodegradable polymer or copolymer. In a more preferred embodiment, the polymer is a poly(lactide-co-glycolide) (hereinafter "PLGA") with a lactide:glycolide ratio of about 1:1 and a molecular weight of about 5,000 Daltons to about 70,000 Daltons. In an even more preferred embodiment, the molecular weight of the PLGA used in the

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present invention has a molecular weight of about 10,000 Daltons.

The amount of EPO, which is contained in a dose of sustained release microparticles, or in an alternate
5 sustained release device, containing biologically active, aggregation-stabilized EPO particles is a therapeutically or prophylactically effective amount, which can be determined by a person of ordinary skill in the art taking into consideration factors such as body weight, condition
10 to be treated, type of polymer used, and release rate from the polymer.

In one embodiment, an EPO sustained release composition will contain from about 0.01% (w/w) to about 50% (w/w) of biologically active, aggregation-stabilized
15 EPO particles. The amount of such EPO particles used will vary depending upon the desired effect of the EPO, the planned release levels, the times at which EPO should be released, and the time span over which the EPO will be released. A preferred range of EPO particle loading is
20 between about 0.1% (w/w) to about 30% (w/w) EPO particles. A more preferred range of EPO particle loading is between about 0.1% (w/w) to about 10% (w/w) EPO particles. The most preferred loading of the biologically active, aggregation-stabilized EPO particles is about 5% (w/w).

25 In yet another embodiment, an EPO sustained release composition also contains a biocompatible metal cation component, which is not contained in the biologically active, aggregation-stabilized EPO particles, but is dispersed within the polymer.

30 A metal cation component, as defined herein, is a component containing at least one kind of multivalent metal cation (having a valence of +2 or more) in a non-dissociated state, dissociated state, or a combination of non-dissociated and dissociated states. Suitable metal
35 cation components include, for instance, metal salts, metal hydroxides, and basic (pH of about 7 or higher) salts of

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weak acids wherein the salt contains a metal cation. It is preferred that the metal cation be divalent.

A metal cation component is biocompatible if it is non-toxic to the recipient, in the quantities used, and
5 also presents no significant deleterious or untoward effects on the recipient's body, such as an immunological reaction at the injection site.

The metal cation component can optionally contain cation species and/or anion species which is contained in
10 an anti-aggregation agent in the EPO particles. The metal cation component acts to modulate the release of the EPO from the polymeric matrix of the sustained release composition and can also enhance the stability of EPO in the composition. In a modulated EPO release, at least one
15 EPO release characteristic, such as the EPO initial release level, the subsequent EPO release levels, duration of release and/or the amount of EPO released, is different from the release characteristics exhibited by EPO being released from a polymeric matrix, wherein the polymeric
20 matrix does not contain a dispersed metal cation component.

A metal cation component used in modulating release typically comprises at least one type of multivalent metal cations. Examples of metal cation components suitable to modulate EPO release, include, or contain, for instance,
25 $\text{Mg}(\text{OH})_2$, MgCO_3 (such as $4\text{MgCO}_3 \cdot \text{Mg}(\text{OH})_2 \cdot 5\text{H}_2\text{O}$), ZnCO_3 (such as $3\text{Zn}(\text{OH})_2 \cdot 2\text{ZnCO}_3$), CaCO_3 , $\text{Zn}_3(\text{C}_6\text{H}_5\text{O}_7)_2$, $\text{Mg}(\text{OAc})_2$, MgSO_4 , $\text{Zn}(\text{OAc})_2$, ZnSO_4 , ZnCl_2 , MgCl_2 and $\text{Mg}_3(\text{C}_6\text{H}_5\text{O}_7)_2$. A suitable ratio of metal cation component-to-polymer is between about 1:99 to about 1:2 by weight. The optimum ratio depends
30 upon the polymer and the metal cation component utilized. A polymeric matrix containing a dispersed metal cation component to modulate the release of a biologically active agent from the polymeric matrix is further described in co-pending U.S. Patent Application No. 08/237,057, filed May
35 3, 1994 and co-pending PCT Patent Application PCT/US95/05511, the teachings of which are incorporated herein by reference in their entirety.

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In yet another embodiment, at least one pore forming agent, such as a water soluble salt, sugar or amino acid, is included in the microparticle to modify the microstructure of the microparticle. The proportion of pore forming agent added to the polymer solution is between about 1% (w/w) to about 30% (w/w). It is preferred that at least one pore forming agent be included in a nonbiodegradable polymeric matrix of the present invention.

An EPO sustained release composition of this invention can be formed into many shapes such as a film, a pellet, a cylinder, a disc or a microparticle. A microparticle, as defined herein, comprises a polymeric component having a diameter of less than about one millimeter and having particles of biologically active, aggregation-stabilized EPO dispersed therein. A microparticle can have a spherical, non-spherical or irregular shape. It is preferred that a microparticle be a microsphere. Typically, the microparticle will be of a size suitable for injection. A preferred size range for microparticles is from about 1 to about 180 microns in diameter, such as for injection through a 23-gauge needle.

In the method of this invention for forming a composition for the sustained release of biologically active, non-aggregated EPO, a suitable amount of biologically active, aggregation-stabilized EPO particles are dispersed within a polymer solution. The EPO particles can be dispersed with the polymer solution by stirring, agitation, sonication or by other known mixing means. The polymer solution, having a dispersion of biologically active, aggregation-stabilized EPO is then solidified, by appropriate means, to form an EPO sustained release composition of this invention.

Alternately, biologically active, aggregation-stabilized EPO particles and a polymer can be mixed into a polymer solvent sequentially, in reverse order, intermittently, separately or through concurrent additions,

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to form a dispersion of the EPO particles in a polymer solution.

A suitable polymer solution contains between about 1% (w/w) and about 30% (w/w) of a suitable biocompatible polymer, wherein the biocompatible polymer is typically dissolved in a suitable polymer solvent. Preferably, a polymer solution contains about 2% (w/v) to about 20% (w/v) polymer. A polymer solution containing 5% to about 10% (w/w) polymer is most preferred.

A suitable polymer solvent, as defined herein, is solvent in which the polymer is soluble but in which the aggregation-stabilized EPO particles are substantially insoluble and non-reactive. Examples of suitable polymer solvents include polar organic liquids, such as methylene chloride, chloroform, ethyl acetate and acetone.

The preparation of EPO sustained release microparticles of the present invention is further described in Example 2.

To prepare biologically active, aggregation-stabilized EPO particles, EPO is mixed in a suitable aqueous solvent with at least one suitable anti-aggregation agent to form a stabilizing mixture, wherein each component of the stabilizing mixture can be in suspension or solution, or a combination thereof.

In forming a stabilizing mixture, the content of anti-aggregation agent is typically between about 10% (w/w) and about 80% (w/w) of the total solids in EPO particles and is preferentially more than about 40% (w/w).

It is understood that the EPO can be in a solid or a dissolved state, prior to being contacted with the anti-aggregation agent. It is also understood that the anti-aggregation agent can be in a solid or a dissolved state, prior to being contacted with the EPO. In a preferred embodiment, an aqueous solution of EPO is mixed with an aqueous solution of the anti-aggregation agent to form a stabilizing mixture.

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The stabilizing mixture is then formed into particles of biologically active, aggregation-stabilized erythropoietin. Any method can be used which is known in the art for forming a solid particle from a mixture of two components wherein one component is temperature sensitive. For example, a solvated stabilizing mixture can be evaporated, lyophilized or freeze-dried. A solid stabilizing mixture can be pressed into pellets.

In a more preferred embodiment, the stabilizing mixture is a solvated mixture which is buffered with an amount of buffer which will maintain pH in a range which will prevent a significant loss of biological activity resulting from pH changes during particle formation (e.g., during lyophilization). Typically, the content of buffer to EPO in a stabilizing mixture is between about 5% (w/w) and about 20% (w/w) of the total solids. Suitable solvents are those in which the EPO and the anti-aggregation agent are each at least slightly soluble, such as in an aqueous sodium bicarbonate buffer or in an aqueous phosphate buffer. For aqueous solvents, it is preferred that water used be either deionized water or water-for-injection.

The stabilizing mixture is usually buffered to a pH between about 4.0 and about 8.0. A preferred pH range is between about 5.0 and about 7.0. Suitable pH conditions are typically achieved through use of an aqueous buffer, such as sodium bicarbonate. More preferred buffers include phosphate buffers, citrate buffers and combinations thereof.

A suitable pH range can be achieved by dialysis with a buffer, by using the buffer as a solvent for the EPO and/or anti-aggregation agent, and by making one or more bulk additions of buffer to the EPO solution before, during, and/or after addition of the anti-aggregation agent.

A solvated stabilizing mixture is then dried, such as by lyophilization, to form particles of biologically active, aggregation-stabilized EPO. The stabilizing mixture

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can be bulk dried or can be divided into smaller volumes which are then dried.

In a preferred embodiment, the stabilizing mixture is micronized, such as by use of an ultrasonic nozzle, frozen
5 and then lyophilized to form biologically active, aggregation-stabilized EPO particles.

In another preferred embodiment, a stabilizing mixture is a buffered solution comprising EPO, ammonium sulfate (about 6 grams to about 8 grams of ammonium sulfate per
10 gram of EPO) and a buffer solution (about 5 grams to about 30 grams of buffer solution per gram of EPO) with a pH between about 5 and about 7. More preferably, the buffer solution is a 5 mM citrate/5 mM phosphate buffer or 5mM phosphate buffer (pH 7).

15 Preferably, particles of aggregation-stabilized EPO are between about 1 to about 6 micrometers in diameter. The EPO particles can be fragmented separately, as described in co-pending U.S. Patent Application No. 08/006,682, filed January 21, 1993, which describes a
20 process for producing small particles of biologically active agents, which is incorporated herein in its entirety by reference. Alternately, the EPO particles can be fragmented after being added to a polymer solution, such as by means of an ultrasonic probe or ultrasonic nozzle.

25 The synthesis of biologically active, aggregation-stabilized EPO particles is further described in Example 1. Additional description of microspheres containing particles of biologically active, aggregation-stabilized EPO and their release pharmacokinetics and
30 pharmacodynamics is provided in Examples 2-5.

In yet another embodiment of the method of this invention, a metal cation component, not contained in EPO particles, is also dispersed within the polymer solution to modulate the release of EPO.

35 It is understood that a metal cation component and the aggregation-stabilized EPO particles can be dispersed into a polymer solution sequentially, in reverse order,

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intermittently, separately or through concurrent additions.

Alternately, a polymer, a metal cation component and the aggregation-stabilized EPO and can be mixed into a polymer solvent sequentially, in reverse order,

5 intermittently, separately or through concurrent additions.

The method for forming a composition for modulating the release of a biologically active agent from a biodegradable polymer is further described in co-pending U.S. Patent Application No. 08/237,057 and co-pending PCT
10 Patent Application PCT/US95/05511.

Further description of microspheres containing particles of aggregation-stabilized EPO and a metal cation component is provided in Example 2.

One suitable method for solidifying a polymer solution
15 to form a polymeric matrix, containing particles of aggregation-stabilized EPO, is the solvent evaporation method described in U.S. Patent No. 3,737,337, issued to Schnoring et al., U.S. Patent No. 3,523,906, issued to Vrancken et al., U.S. Patent No. 3,691,090, issued to
20 Kitajima et al., or U.S. Patent No. 4,389,330, issued to Tice et al, which are incorporated herein in their entirety by reference. Solvent evaporation can be used to form microparticles or other shaped EPO sustained release devices.

25 In the solvent evaporation method, a polymer solution containing an aggregation-stabilized EPO particle dispersion, is mixed in or agitated with a continuous phase, in which the polymer solvent is partially miscible, to form an emulsion. The continuous phase is usually an
30 aqueous solvent. Emulsifiers are often included in the continuous phase to stabilize the emulsion. The polymer solvent is then evaporated over a period of several hours or more, thereby solidifying the polymer to form a polymeric matrix having a dispersion of aggregation-
35 stabilized EPO particles contained therein. In this method, care must be taken not to heat the polymer solution to a temperature at which denaturing of the EPO in the EPO

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particles could occur. Further discussion of the high level of biological activity, typically > 98%, maintained in the microparticles of the present invention is provided in Example 2.

5 Another suitable method for solidifying a polymer solution to form a polymeric matrix, containing particles of aggregation-stabilized EPO, is the phase separation method described in U.S. Patent No. 4,675,800, which is incorporated herein in its entirety by reference. In this
10 method, polymer within a polymer solution containing aggregation-stabilized EPO particles is precipitated around the EPO particles by the addition of a polymer non-solvent to the polymer solution to form an emulsion, wherein the polymer non-solvent is immiscible with the polymer solvent.

15 A preferred method for forming aggregation-stabilized EPO microparticles from a polymer solution uses rapid freezing and solvent extraction, as described in U.S. Patent No. 5,019,400, issued to Gombotz et al. and co-pending U.S. Patent Application No. 08/443,726, filed
20 May 18, 1995, the teachings of which are incorporated herein by reference in their entirety. This method of microsphere formation, as compared to other methods, such as phase separation, additionally reduces the amount of EPO required to produce a sustained release composition with a
25 specific EPO content. Also see Example 2 for additional descriptions of microparticle formulations by this method.

In this method, the polymer solution, containing the EPO particle dispersion, is processed to create droplets, wherein at least a significant portion of the droplets
30 contain polymer solution and biologically active, aggregation-stabilized EPO particles. These droplets are then frozen by means suitable to form microparticles. Examples of means for processing the polymer solution dispersion to form droplets include directing the
35 dispersion through an ultrasonic nozzle, pressure nozzle, Rayleigh jet, or by other known means for creating droplets from a solution.

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- Means suitable for freezing droplets to form microparticles include directing the droplets into or near a liquified gas, such as liquid argon and liquid nitrogen to form frozen microdroplets which are then separated from the liquid gas. The frozen microdroplets are then exposed to a liquid non-solvent, such as ethanol, or ethanol mixed with hexane or pentane. The solvent in the frozen microdroplets is extracted as a solid and/or liquid into the non-solvent to form biologically active, aggregation-stabilized EPO containing microparticles. Mixing ethanol with other non-solvents, such as hexane or pentane, can increase the rate of solvent extraction, above that achieved by ethanol alone, from certain polymers, such as poly(lactide-co-glycolide) polymers.
- A wide range of sizes of EPO sustained release microparticles can be made by varying the droplet size, for example, by changing the ultrasonic nozzle diameter. If very large microparticles are desired, the microparticles can be extruded through a syringe directly into the cold liquid. Increasing the viscosity of the polymer solution can also increase microparticle size. For example, the size of the microparticles produced by this process can range from greater than 1000 to 1 micrometers or less in diameter.
- Yet another method of forming an EPO sustained release composition from a polymer solution, includes film casting, such as in a mold, to form a film or a shape. For instance, after putting the polymer solution containing a dispersion of aggregation-stabilized EPO particles into a mold, the polymer solvent is then removed by means known in the art, or the temperature of the polymer solution is reduced, until a film or shape, with a consistent dry weight, is obtained. Film casting of a polymer solution, containing a biologically active agent, is further described in co-pending U.S. Patent Application No. 08/237,057.

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The method of this invention for forming a biologically active EPO sustained release composition can also be used to form a sustained release composition of other biologically active agents which are soluble in aqueous solutions and at sufficiently high concentrations in solution, have a tendency to aggregate to form biologically inactive polymers.

It is believed that the release of the EPO can occur by two different mechanisms. The EPO can be released by diffusion through aqueous filled channels generated in the polymeric matrix, such as by the dissolution of the EPO or by voids created by the removal of the polymer's solvent during the synthesis of the sustained release composition. The rate of polymer hydrolysis may be increased by non-neutral pH's. Therefore, an acidic or a basic excipient can be added to the polymer solution, used to form the microparticle, to alter the polymer erosion rate.

The second mechanism is the release of EPO due to degradation of the polymer. The rate of degradation can be controlled by changing polymer properties that influence the rate of hydration of the polymer. These properties include, for instance, the ratio of different monomers, such as lactide and glycolide, comprising a polymer; the use of the L-isomer of a monomer instead of a racemic mixture; the end groups and the molecular weight of the polymer. These properties can affect hydrophilicity and crystallinity, which control the rate of hydration of the polymer. Hydrophilic excipients such as salts, carbohydrates and surfactants can also be incorporated to increase hydration and which can alter the rate of erosion of the polymer.

By altering the properties of the polymer, the contributions of diffusion and/or polymer degradation to EPO release can be controlled. For example, decreasing the molecular weight of the polymer, using an unblocked polymer instead of a blocked polymer, or increasing the glycolide content of a poly(lactide-co-glycolide) polymer can enhance

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the hydrolysis of the polymer and thus provide an increased EPO release from polymer erosion. The effects of different molecular weight unblocked PLGA polymers are further described in Example 5.

- 5 The sustained release composition of this invention can be administered to a human, or other animal, by injection, implantation (e.g, subcutaneously, intramuscularly, intraperitoneally, intracranially, intravaginally and intradermally), administration to
- 10 mucosal membranes (e.g., intranasally or by means of a suppository), or *in situ* delivery (e.g. by enema or aerosol spray) to provide the desired dosage of EPO based on the known parameters for treatment with EPO of the various medical conditions.
- 15 The invention will now be further and specifically described by the following examples.

Example 1

Preparation of Biologically Active, Aggregation-Stabilized Erythropoietin

- 20 Erythropoietin was derived as described in U.S. Patent No. 4,703,008. The EPO was dissolved in deionized water to form an aqueous solution having a concentration of approximately 1 mg/ml. Different samples of the EPO solution were then dialyzed against three changes of the
- 25 appropriate formulation buffer (i.e., 5mM phosphate buffer (pH 7), 5 mM citrate buffer (pH 7), 5 mM citrate/5mM phosphate buffer (pH 7) or 10 mM bicarbonate buffer (pH 7)).

- 30 Following dialysis, the concentration of EPO in the dialyzed solutions was verified to be approximately 1 mg/ml as determined by measuring absorbance at 280 nm ($\epsilon = 1.345 \text{ L gm}^{-1} \text{ cm}^{-1}$).

- 35 Portions of the dialyzed EPO solutions were then mixed with concentrated solutions of candidate anti-aggregation agents (i.e., ammonium sulfate, zinc acetate,

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mannitol/sucrose or mannitol/maltose) to form the EPO formulations provided in Table I below. The candidate anti-aggregation agent solutions also possibly contained additional excipients (i.e, inulin, glycine and TWEEN 20™
5 surfactant).

The anti-aggregation agent solutions were separately prepared in the same buffers used to dialyze the EPO solutions to which they were subsequently added.

Approximate volumes of each anti-aggregation agent
10 solution and of additional buffer were added to a 50 ml polypropylene tube to achieve the desired concentrations for the formulations (described in Table I). Each dialyzed EPO solution was then added to the appropriate anti-aggregation agent solution and then the solutions were
15 mixed by gentle inversion.

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Table I

<u>Formulations</u> (wt %)		<u>Am1</u>	<u>Am4</u>	<u>Am7</u>	<u>Ma1</u>	<u>Ma3</u>	<u>Ma4</u>	<u>Zn1</u>	<u>Zn6</u>
	EPO	10.0	10.1	9.9	10.0	10.0	10.0	10.0	10.0
5	Ammonium Sulfate	66.8	64.7	79.1	0.0	0.0	0.0	0.0	0.0
	Zinc Acetate	0.0	0.0	0.0	0.0	0.0	0.0	76.9	76.9
	Mannitol	0.0	0.0	0.0	62.5	62.5	72.5	0.0	0.0
	Sucrose	0.0	0.0	0.0	10.0	0.0	10.0	0.0	0.0
10	Maltose	0.0	0.0	0.0	0.0	10.0	0.0	0.0	0.0
	5mM Citrate Buffer (pH 7)	0.0	15.1	0.0	0.0	0.0	0.0	0.0	0.0
	5mM Phosphate Buffer (pH 7)	0.0	0.0	10.0	7.5	7.5	7.5	0.0	0.0
15	5mM Citrate/ 5mM Phosphate Buffer (pH 7)	22.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	10mM Bicarbonate Buffer (pH 7)	0.0	0.0	0.0	0.0	0.0	0.0	13.1	12.1
20	Inulin	1.1	10.1	1.0	0.0	0.0	0.0	0.0	0.0
	Glycine	0.0	0.0	0.0	10.0	10.0	0.0	0.0	0.0
	TWEEN 20™ Surfactant	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.0

- 25 Each of the formulated EPO solutions were aspirated into a 60 ml plastic syringe fitted with Teflon tubing and then atomized through an ultrasonic nozzle (Type V1A; Sonics and Materials, Inc., Danbury, CT) into polypropylene containers containing liquid nitrogen, making sure that the
- 30 atomized material was fully submerged at all times, to form frozen particles. The container was kept at -80°C until the liquid nitrogen was entirely evaporated. The frozen

1000000/ml
↓
1070
1000000/ml

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particles, which contained biologically active, aggregation-stabilized EPO, were transferred into glass beakers then lyophilized to form biologically active, aggregation-stabilized EPO particles. The EPO particles
5 were removed from the lyophilizer under an atmosphere of dry nitrogen, handled in a low humidity environment, and stored desiccated at -80°C.

Example 2

Preparation of PLGA Microspheres Containing 10 Biologically Active, Aggregation-Stabilized Erythropoietin

Microspheres containing the aggregation-stabilized EPO formulations of Example 1 were prepared from unblocked (50:50; MW 10,000 Daltons) PLGA, obtained from Boehringer
15 Ingelheim Chemicals, Inc., Montvale, NJ, or blocked (50:50; MW 10,000 Daltons) PLGA obtained from Birmingham Polymers, Inc., Birmingham, AL.

In addition, microspheres, containing the Am7 formulation of aggregation-stabilized EPO particles, were prepared from unblocked (50:50) PLGA with a molecular
20 weight of approximately 31,000 Daltons or 45,000 Daltons, (Boehringer Ingelheim Chemicals, Inc., Montvale, NJ).

The method described in Gombotz et al. (U.S. Patent No. 5,019,400) was used to encapsulate the aggregation-stabilized EPO particles of Example 1 in PLGA. In each
25 case, polymer was dissolved in 5.1 ml of methylene chloride to form a polymer solution. Magnesium carbonate, or zinc carbonate, was sieved through a 38 micrometer sieve and was then added to the polymer solution to a final concentration of 10% w/vol. The polymer/salt suspension was subsequently
30 combined with 30 mg of aggregation-stabilized EPO particles.

The polymer solution, containing suspended salt and EPO particles, was placed in an ice-water bath and sonicated using an ultrasonic probe (Virtis Co., Gardiner,
35 NY) to reduce the protein particle size to approximately 2

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to 3 micrometers in diameter and to form a dispersion of EPO particles within the polymer solution.

A frozen bed of ethanol was prepared in a polypropylene tub by surrounding the tub with liquid nitrogen, then covering the frozen ethanol with additional liquid nitrogen. The EPO/polymer suspension was then pumped at a rate of 1-2 ml/min from a syringe by a syringe pump (Orion Research Inc., Boston, MA) into an ultrasonic nozzle placed above the container of frozen ethanol and liquid nitrogen. The EPO/polymer suspension was atomized into droplets which froze upon contact with the liquid nitrogen and formed microspheres which sank to the surface to the frozen ethanol.

The container was placed at -80°C to evaporate the liquid nitrogen and allow the ethanol to melt. As the ethanol thawed, the microspheres sank into the ethanol and the methylene chloride was extracted from the microspheres. After 24 hours, additional ethanol, which was prechilled to -80°C , was added to the container. Two days after the microspheres were prepared, the ethanol/microsphere slurry was filtered through a 0.65 micron Durapore[®] membrane (Millipore, Bedford, MA) using a precooled filter apparatus. The filtration was performed in a glove box that was preflushed with nitrogen gas. The filtered microspheres were then lyophilized (on a shelf that had been precooled to -40°C) until dry.

The immunoreactivity of the EPO in these sustained release microspheres was subsequently determined by extracting protein and analyzing by radioimmunoassay (RIA) (Incstar: Stillwater, MN). To extract the EPO from the microspheres, approximately 10 mg of microspheres were placed in a tube with 250 μl of methylene chloride. The samples were vortexed for 10 to 20 seconds and left at room temperature for 5 minutes to dissolve the polymer. A sample of acetone (750 μl) was added, vortexed for an additional 10 seconds, and centrifuged at 14,000 rpm for 30 seconds at 4°C to pellet the EPO. The supernatant was

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removed and the methylene chloride and acetone steps were repeated twice more. Samples were dried in a lyophilizer or vacuum oven for 14-18 hours at room temperature. The EPO pellet was reconstituted in 1 ml HEPES buffer by
5 vortexing for about 10 seconds, then standing at room temperature for about 1 hour until completely dissolved. The extracted EPO was diluted in buffer (8.1 mM Na_2HPO_4 , 1.5 mM KH_2PO_4 , 400 mM NaCl, pH 7.5) to a concentration of approximately 25 $\mu\text{g/ml}$ for analysis.

10 The immunoreactivity of the EPO was found to be 121,000 \pm 5000 units per mg of EPO. This specific activity is comparable to the range obtained for bulk EPO (130,000-140,000 units per mg of EPO, thus showing an insignificant reduction of EPO activity due to the method of forming the
15 sustained release compositions of the present invention. Monomer content was found to be greater than 98% for all microspheres.

The microspheres containing Am1 and Am7 EPO particles were also assayed for EPO dimer, by size exclusion
20 chromatography (SEC), and for high molecular weight EPO aggregates by SDS-PAGE/ Western blot analysis. No EPO dimer or high molecular weight aggregates were detected.

Example 3

In Vitro Release of EPO From

25 Aggregation-Stabilized Particles Within PLGA Microspheres

The in vitro release kinetics of EPO from aggregation-stabilized particles within PLGA microspheres were assessed in HEPES buffer (75 mM HEPES, 115 mM NaCl, 0.1% (by volume) TWEEN 20TM, 0.1% (by weight) sodium azide titrated to pH
30 7.4 with NaOH) or in HEPES buffer containing 2% or 20% sheep serum. The serum containing buffer is 80% by volume above plus 20% by volume sheep serum. The studies were conducted by suspending 8-10 mg of microspheres in 1-5[?]

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ml of buffer at 37°C. At specified time points, the buffer was removed in full and replaced with fresh buffer.

In samples incubated in HEPES buffer, the releases over time of EPO monomer (biologically active EPO) and of
5 EPO aggregates (biologically inactive EPO) were determined by size exclusion chromatography (SEC). The results of the SEC analyses upon *in vitro* release kinetics in HEPES buffer of various microspheres, wherein the microspheres were a) unblocked PLGA (MW 10,000 Daltons) microspheres containing
10 formulations Am1 or Am7, and b) blocked PLGA (MW 10,000 Daltons) microspheres containing Zn1, are provided in Figures 1, 2 and 3, respectively. Figures 1 and 2 show the EPO released from formulations containing ammonium sulfate as an anti-aggregation agent was almost all monomeric EPO
15 over the length of the release period.

Figure 3 shows the EPO released from a formulation containing zinc acetate, as an anti-aggregation agent, contained significant levels of aggregate which increased substantially over the length of the release period.

20 The results of the SEC and RIA analyses upon *in vitro* release kinetics in HEPES buffer, and in HEPES/serum, of various microspheres (all in 10,000 Dalton PLGA) which contained different EPO formulations of Example 1 are provided in Table II. The initial burst and release rate
25 were determined in the HEPES/serum test by RIA. The integrity of the released EPO was assessed in HEPES buffer by SEC.

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Table II

	<u>Formula</u>	<u>EPO</u> <u>Load</u> (%)	<u>Polymer/</u> <u>Salt</u>	<u>Aggregate</u> <u>Released</u> (% init.load)	<u>Initial</u> <u>Burst</u> (%)	<u>Average</u> <u>Release</u> (%/day)	<u>Release</u> <u>Duration</u> (days)
5	Zn1	10	Blocked/ 10% MgCO ₃	12	66	1.2	14
	Zn1	10	Blocked/ 10% ZnCO ₃	22	46	1.7	28
	Zn6	10	Blocked/ 10% ZnCO ₃	37	32	1.6	28
10	Am1	5	Unblocked/ 10% MgCO ₃	1	39	1.4	21
	Am1	10	Blocked/ 10% MgCO ₃	2	71	0.3	3
15	Am4	5	Unblocked/ 10% MgCO ₃	1	29	1.1	21
	Am4	5	Unblocked/ none	1	35	0.9	28
	Ma1	5	Unblocked/ 10% MgCO ₃	1	44	1.8	24
20	Ma3	10	Unblocked/ 10% MgCO ₃	1	71	1.3	21
	Ma4	10	Blocked/ 10% ZnCO ₃	1	77	0.6	3

25 These analyses show that the addition of suitable
 anti-aggregation agents significantly reduced the
 aggregation of EPO over the release periods. These
 analyses also demonstrated that the addition of a metal
 cation component (e.g., salt) to the polymer, as well as
 30 the selection of the type of polymer (i.e., blocked or
 unblocked) significantly affected the initial burst level
 and the duration of release.

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Example 4Integrity of EPO Released from Ammonium SulfateAggregation-Stabilized EPO ParticlesWithin PLGA Microspheres

5 The purpose of the experiment was to determine the integrity of EPO released from microshperes having varying concentrations of ammonium sulfate.

Aggregation-stabilized EPO formulations comparable to AM7, except having 10%, 20%, or 40% ammonium sulfate, were
10 prepared as described in Example 1. The eliminated ammonium sulfate was replaced with sodium chloride or sucrose such that the total weight of ammonium sulfate and sodium chloride or sucrose was 79%.

The percent monomeric and aggregate EPO were
15 determined after 35 days and 42 days release in vitro. The AM7 formulation, as well as the 40% ammonium sulfate/NaCl formulation produced 3-4% aggregates at both time points, whereas the 10% and 20% ammonium sulfate/NaCl formulations produced 5-6% aggregates. Mannitol formulations produced
20 results similar to the 10% and 20% ammonium sulfate formulations.

In the case where ammonium sulfate was replaced with sucrose, there was not sufficient drug released from the 40% ammonium sulfate formulation to quantitate. The 10%
25 and 20% ammonium sulfate formulations with sucrose, like their sodium chloride counterparts, showed more aggregates (6-9%) than were observed with the AM7 formulation.

Example 5Effect of Co-Administered Cyclosporin and Hydrocortisoneon In Vivo Pharmacokinetics of Erythropoietin
30

Male Sprague-Dawley rats, weighing 400 ± 50 g (S.D.), were used as the animal model. The rats were not fasted before the experiments and subsequently were fed with a standard diet, an iron supplement, and allowed free access

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to water. Iron dextran (Sigma Co., St. Louis, MO) 5 mg/kg was injected intraperitoneally twice a week.

These experiments utilized the method described in co-
pending U.S. Patent Application No. 08/480,813, filed June
5 7, 1995, Attorney Docket No. ACT95-03, of suppressing
antibody production in the test animals in response to the
EPO released (or injected) to obtain accurate profiles of
serum EPO levels. Therein, antibody production is
suppressed by administration of cyclosporin A and
10 hydrocortisone to the test animal.

A purpose of the first experiment was to compare the
in vivo pharmacodynamic effects of EPO released from
sustained release microspheres to EPO injected
subcutaneously as a bolus, specifically upon serum
15 reticulocyte profiles. Two groups of, three rats were
injected subcutaneously in the interscapular region on day
0 with 10,000 units of RMAM7 EPO microspheres (unblocked
10K PLGA containing 10% MgCO₃ and 5% Am7) and subsequently
on day 28 with a 2,000 unit bolus of aqueous EPO. The
20 control group did not receive the cyclosporin A/
hydrocortisone therapy, which the test group did receive.

Blood samples were taken from the tail vein of each
rat at 1, 3, 4, 8, 10, 14, 16, 20, 24, 28, 30 or 31, 32 and
36 hours after injection. Additional blood samples were
25 then taken approximately once a day for the following 4-5
days.

Blood reticulocyte levels were counted for each blood
sample. The results are provided in Figure 4. Figure 4
shows higher reticulocyte counts in immunosuppressed rats
30 in response to both the EPO microspheres and the EPO bolus.
The non-immunosuppressed rats (control group) showed lower
reticulocyte levels due to antibody formation resulting
from the immune systems' responses to EPO. This is
particularly shown by the lack of a significant increase in
35 reticulocyte levels in the control group after receiving
the EPO bolus on day 28.

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Figure 4 also shows that injection with sustained release microspheres resulted in a longer period of elevated serum reticulocyte levels than did a bolus of EPO.

A purpose of the second experiment was to compare the in vivo pharmacokinetic and pharmacodynamic effects of EPO released from various sustained release microspheres.

The rats in each of four groups rats (N=3) were injected subcutaneously in the interscapular region with one of four of the following formulations of microspheres:

10	RMAm1	Unblocked 10K PLGA/10% MgCO ₃ /5% Am1
	RMMal	Unblocked 10K PLGA 10% MgCO ₃ /5% Mal
	PZZn1	Blocked 10K PLGA/10% ZnCO ₃ /5% Zn1
	RMAm7	Unblocked 10K PLGA/10% MgCO ₃ /5% Am7

Each rat received between 10,000 to 12,000 units per animal. Each rats was also given daily an intraperitoneal injection of 10 mg cyclosporin A (Sandimmune® Injection, Sandoz, East Hanover, NJ) and 5 mg hydrocortisone (Spectrum Co., Gardena, CA) in 0.5 ml sterilized saline for injection (USP) for days 0 to 14 and then injections twice a week for days 15 to 28. These injections were to suppress the response of the rats' immune systems to EPO.

Blood samples were taken from the tail vein of each rat at 1, 2, 4, 8, 10 (optionally), 24, 36 and 48 hours after injection. Additional blood samples were then taken approximately once a day for the following 4-5 days. The EPO concentration in the rat serum samples was determined using by ELISA. In addition, blood reticulocyte levels were counted.

Serum EPO and blood reticulocyte profiles for these formulations are provided in Figures 5 and 6. EPO levels remained above baseline in these animals for approximately 14 days, showing a sustained release of biologically active EPO. Elevated reticulocyte levels were observed for about 17 days. Further, the response of immature and total

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reticulocyte levels were proportional and not significantly different from each other following EPO treatment.

Example 6

Effect of Polymer Molecular Weight on EPO Release in Rats

5 Rats (N=3) in three test groups were injected, as described in Example 4, with the microspheres of Example 2, having the Am7 EPO particle formulation, in different molecular weight PLGA polymers (10,000 Daltons, 31,000 Daltons or 45,000 Daltons). The dose for each rat was
10 about 10,000 units.

The purpose of the test was to determine the effects of polymer molecular weight upon sustained EPO release levels and release period.

Blood samples were taken from the tail vein of each
15 rat on days 3, 7, 10, 14, 17, 21, 24 and 28 after injection. The EPO concentration in the rat serum samples was determined using by ELISA. The results are provided in Figure 7. Figure 7 shows that substantially decreasing polymer molecular weight increases the rate of EPO release
20 and shortens the release period in unblocked PLGA containing 2% $MgCO_3$.

EQUIVALENTS

Those skilled in the art will recognize, or be able to ascertain, using no more than routine experimentation, many
25 equivalents to specific embodiments of the invention described specifically herein. Such equivalents are intended to be encompassed in the scope of the following claims.

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CLAIMS

The invention claimed is:

1. A composition for the sustained release of
biologically active, non-aggregated erythropoietin
5 from a polymeric matrix comprising:
 - a) a biocompatible polymer;
 - b) particles of erythropoietin, wherein said
erythropoietin is stabilized against aggregation
with a salt which reduces the solubility of
10 erythropoietin in aqueous solutions.
2. The composition of Claim 1 wherein the salt comprises
a cation selected from the group consisting of Mg^{+2} ,
 Li^{+2} , Na^{+} , K^{+} , NH_4^{+} and combinations thereof.
3. The composition of Claim 1 or 2 wherein the salt
15 comprises an anion selected from the group consisting
of SO_4^{-2} , HPO_4^{-2} , acetate, citrate, tartrate, Cl^{-} ,
 NO_3^{-} , ClO_3^{-} , I^{-} , ClO_4^{-} , SCN^{-} and combinations thereof.
4. The composition of Claim 1 wherein the salt is
ammonium sulfate.
- 20 5. The composition of any one of the preceding claims
wherein the salt content is between about 10% and 80%
by weight or at least about 40% by weight total solids
in the erythropoietin particles.
6. The composition of any one of the preceding claims
25 wherein aggregation of erythropoietin is maintained
below about 5% of initial dose, preferably below about
2%.

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7. The composition of any one of the preceding claims wherein said polymer is selected from the group consisting of poly(lactide), poly(glycolide), poly(lactide-co-glycolide,) poly(lactic acid),
5 poly(glycolic acid), polycaprolactone, polycarbonate, polyesteramide, polyanhydride, poly(amino acid), polyorthoester, polycyanoacrylate, poly(dioxanone), poly(alkylene oxalate), polyurethane, blends and copolymers thereof.
- 10 8. The composition of Claim 7 wherein said polymer is selected from the group consisting of poly(lactide), poly(glycolide) and poly(lactide-co-glycolide).
9. The composition of any one of the preceding claims wherein the particles of erythropoietin are present in
15 said polymer at a concentration between about 0.1% and about 30% by weight, preferably between about 0.1% and about 10% by weight.
10. The composition of Claim 9 wherein the particles of erythropoietin are present in said polymer at a
20 concentration of about 5% by weight.
11. The composition of any one of the preceding claims wherein the erythropoietin is lyophilized in the presence of a buffer maintained at a pH between about 4.0 and about 8.0, preferably between about 5.0 and
25 about 7.0.
12. The composition of Claim 11 wherein the buffer is selected from phosphate and citrate buffers.
13. The composition of any one of the preceding claims wherein the biocompatible polymer further comprises a
30 metal cation component.

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14. The composition of Claim 13 wherein the metal cation of the metal cation component is a multivalent ion, such as Zn^{+2} , Mg^{+2} and Ca^{+2} .
- 5 15. The composition of Claim 14 wherein the metal cation component is selected from the group consisting of $\text{Mg}(\text{OH})_2$, MgCO_3 , ZnCO_3 , CaCO_3 , $\text{Zn}_3(\text{C}_6\text{H}_5\text{O}_7)_2$, $\text{Mg}(\text{OAc})_2$, MgSO_4 , $\text{Zn}(\text{OAc})_2$, ZnSO_4 , ZnCl_2 , MgCl_2 and $\text{Mg}_3(\text{C}_6\text{H}_5\text{O}_7)_2$, preferably MgCO_3 .
- 10 16. The composition of Claim 14 or 15 wherein the ratio of the metal cation component to polymer is between about 1:99 to about 1:2 by weight.
17. The composition of Claim 16 wherein the metal cation component is magnesium carbonate at a concentration of about 10% by weight of the total composition.
- 15 18. The composition of any one of the preceding claims for use in the preparation of a medicament for therapy.
19. Biologically-active, aggregation stabilized erythropoietin, comprising a lyophilized solution of erythropoietin, ammonium sulfate and a buffer having a
20 pH of between about 4 and 8, preferably between about 5 and 7.
20. The erythropoietin of Claim 19 wherein the content of ammonium sulfate is about 10% and 80% by weight or at least about 40% by weight total solids.
- 25 21. The erythropoietin of Claim 20 wherein the buffer is a phosphate or citrate buffer.
22. The erythropoietin of any one of Claims 19-21 for use in the manufacture of a sustained release composition.

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23. A method for preparing particles of biologically-active, aggregation-stabilized erythropoietin, comprising the steps of:
- 5 a) mixing an anti-aggregation agent, biologically active erythropoietin and a buffer at a pH of between about 4 and 8, preferably between about 5 and 7, to form an aggregation-stabilizing mixture; and
- 10 b) lyophilizing said mixture to form biologically active, aggregation-stabilized erythropoietin.
24. The method of Claim 23 wherein the anti-aggregation agent is ammonium sulfate and wherein the ammonium sulfate is added at a concentration between about 10% and 80% by weight or at least about 40% by weight
- 15 total solids.
25. The method of Claim 23 wherein the buffer is a phosphate or citrate buffer.

1/4

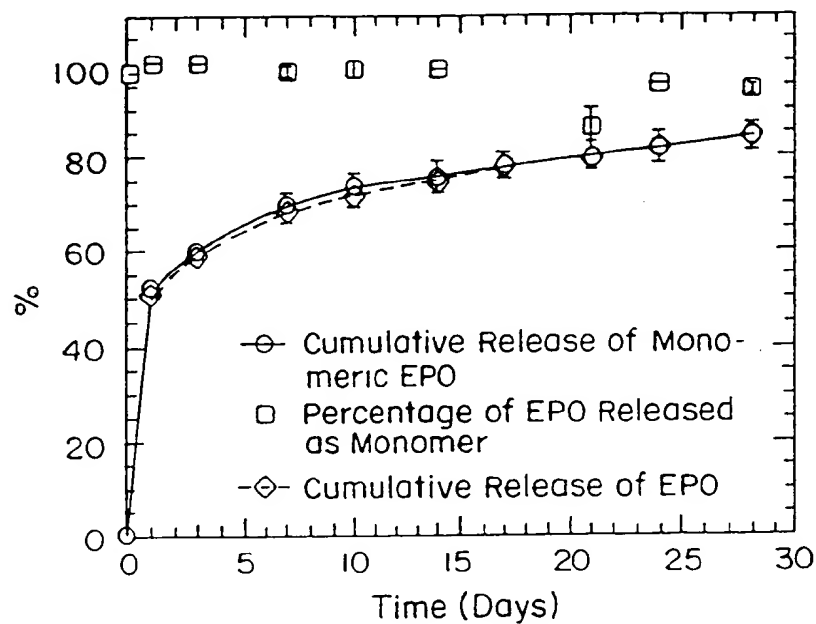


FIG. 1

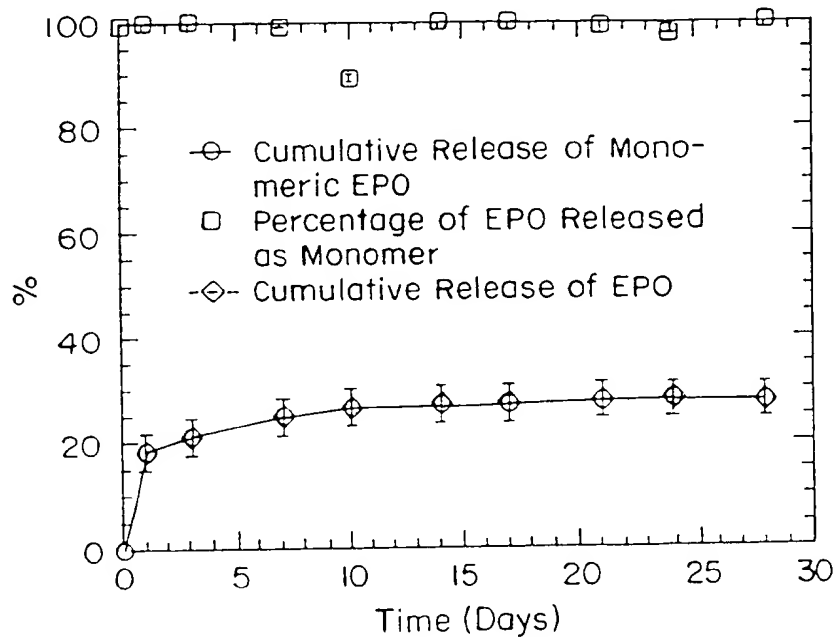


FIG. 2

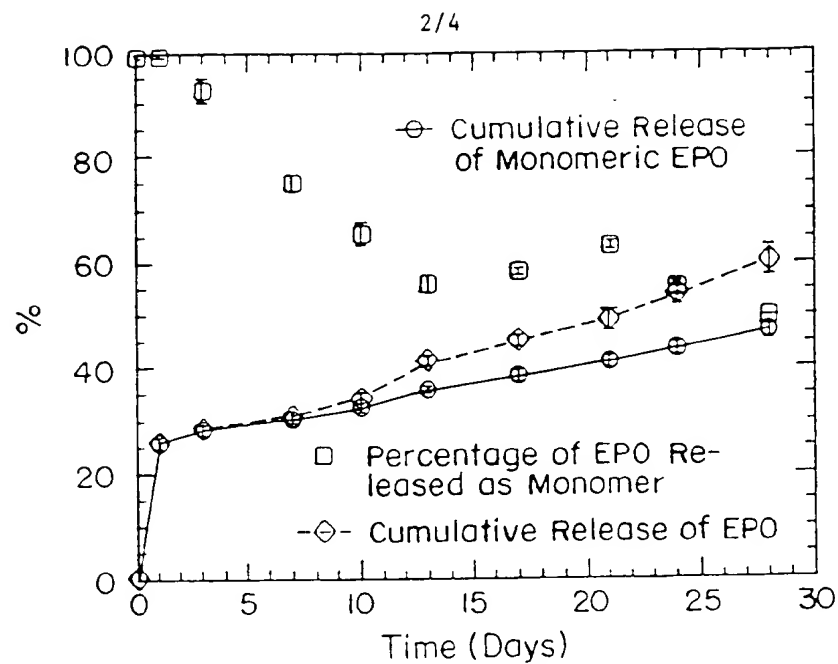


FIG. 3

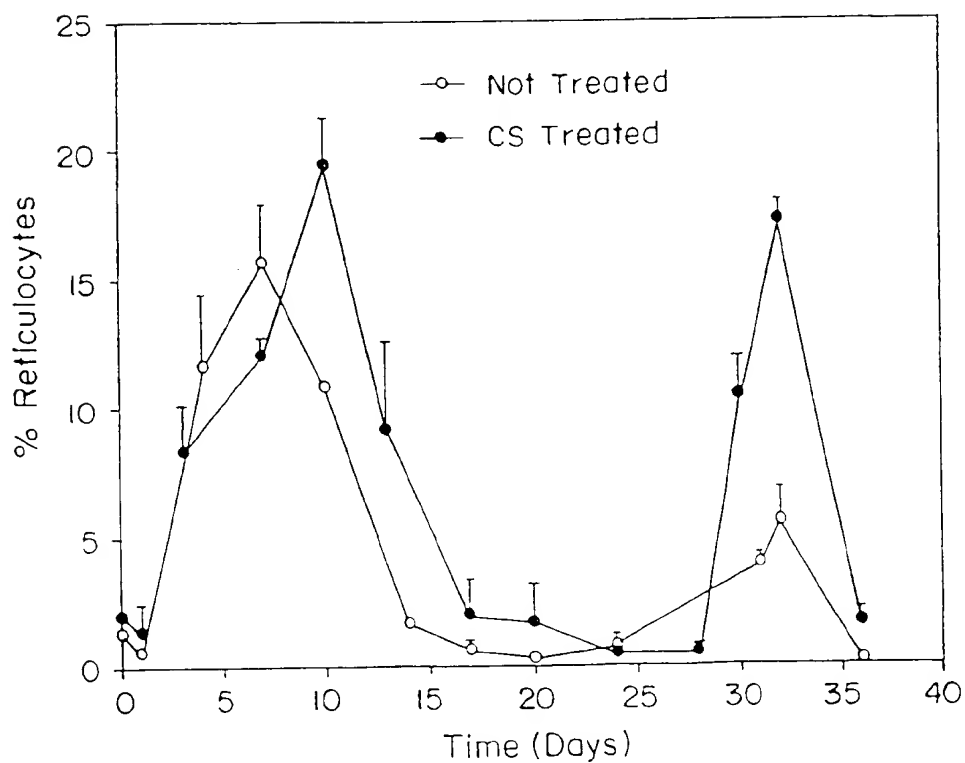


FIG. 4

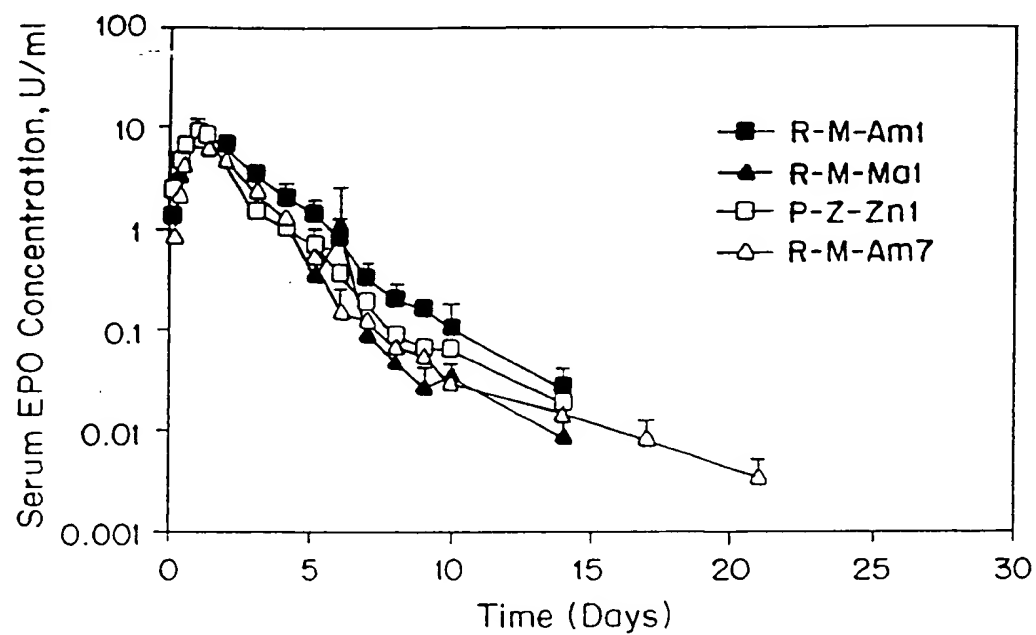


FIG. 5

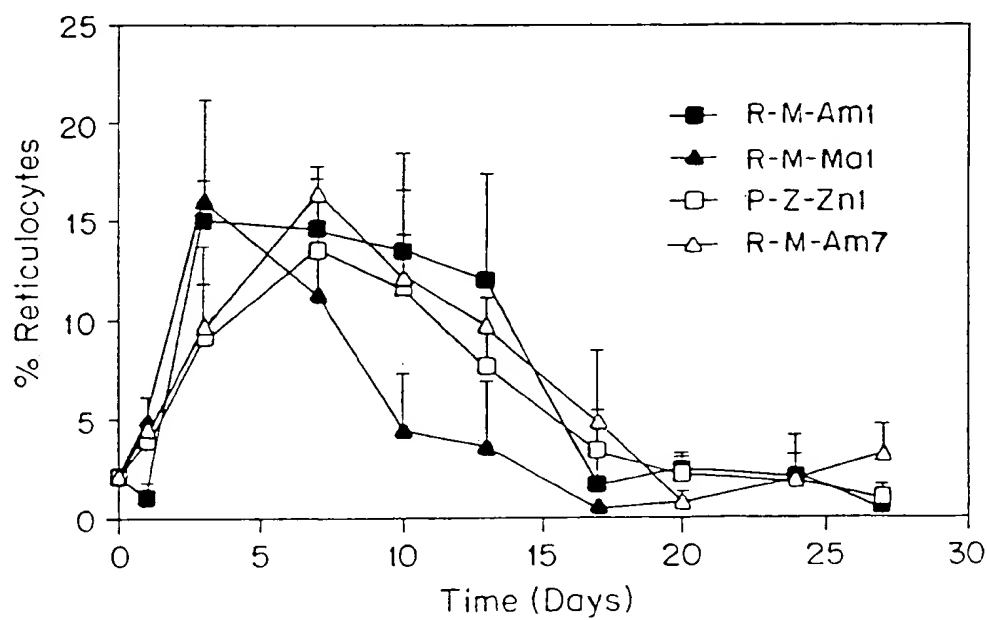


FIG. 6

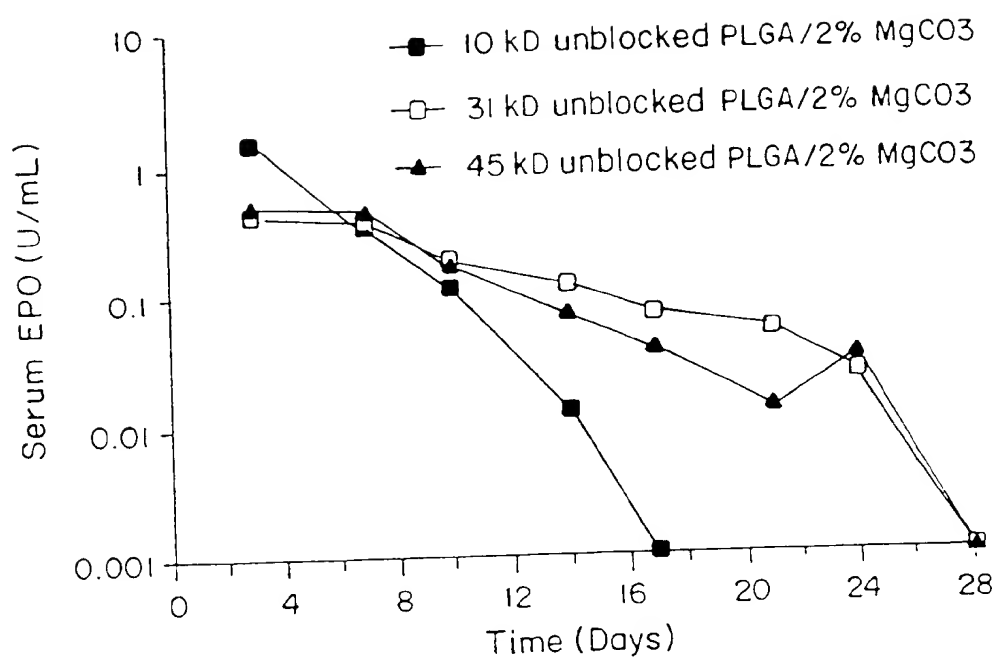


FIG. 7

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INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

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(21) International Application Number: PCT/US96/08474 (22) International Filing Date: 3 June 1996 (03.06.96) (30) Priority Data: 08/478,502 7 June 1995 (07.06.95) US (71) Applicant: ALKERMES CONTROLLED THERAPEUTICS, INC. [US/US]; 4th floor, 64 Sidney Street, Cambridge, MA 02139 (US). (72) Inventors: ZALE, Stephen, E.; 117 Spring Street, Hopkinton, MA 01748 (US). BURKE, Paul, A.; 26 Jackson Road, Medford, MA 02155 (US). BERNSTEIN, Howard; 929 Massachusetts Avenue No. 2B, Cambridge, MA 02139 (US). BRICKNER, Avram; 55 Kenwood Street, Brookline, MA 02146 (US). (74) Agents: BROOK, David, E. et al.; Hamilton, Brook, Smith & Reynolds, Two Militia Drive, Lexington, MA 02173 (US).	(81) Designated States: AL, AM, AT, AU, AZ, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IL, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, UZ, VN, ARIPO patent (KE, LS, MW, SD, SZ, UG), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG). Published <i>With international search report.</i> <i>Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i> (88) Date of publication of the international search report: 23 January 1997 (23.01.97)	
(54) Title: COMPOSITION FOR SUSTAINED RELEASE OF NON-AGGREGATED ERYTHROPOIETIN		
(57) Abstract <p>A composition, and methods of forming and using said composition, for the sustained release of non-aggregated, biologically active, erythropoietin (EPO). The sustained release composition of this invention comprises a polymeric matrix of a biocompatible polymer and particles of biologically active, aggregation-stabilized EPO, wherein said particles are dispersed within the biocompatible polymer. The method of the invention for producing a composition for the sustained release of biologically active EPO, includes dissolving a biocompatible polymer in a polymer solvent to form a polymer solution, dispersing particles of biologically active, aggregation-stabilized EPO in the polymer solution, and then solidifying the polymer to form a polymeric matrix containing a dispersion of said EPO particles. The method for using a composition of the invention is a method for providing a therapeutically effective blood level of biologically active, non-aggregated erythropoietin in a subject for a sustained period. In this method, a subject is administered an effective dose of the sustained release composition of the present invention.</p>		

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INTERNATIONAL SEARCH REPORT

International Application No
PCT/US 96/08474

A. CLASSIFICATION OF SUBJECT MATTER
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According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 6 A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO,A,93 25221 (ALKERMES CONTROLLED THERAPEUTICS) 23 December 1993 see claims 10,1417,19,20 see page 5, line 25 - line 26 see page 14, paragraph 3 - page 15, paragraph 2 see page 26; example 4	1-3,5, 7-10,18
P,X	WO,A,96 07399 (TAKEDA CHEMICAL INDUSTRIES) 14 March 1996 see claims 1-5,11,16,17 see page 7, paragraph 2 see page 8, paragraph 2 see page 9, paragraph 8 - page 10, paragraph 2 see page 11, paragraph 5 - page 12, paragraph 1 see page 26; example 3	1-3, 7-10,18

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Date of the actual completion of the international search

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